





# Laboratory Simulations of Aqueous Reactions in Lunar Polar Regions

Chengzheng Yong, A. Wang, Y. C. Yan, B. L. Jolliff

Washington University in St. Louis

McDonnel Center for the Space Sciences

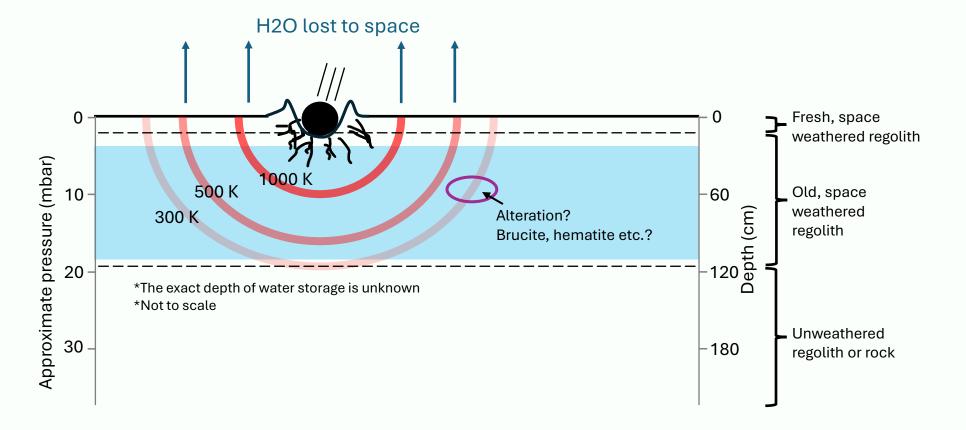




Earth, Environmental, & Planetary Sciences

## Background

- We might expect aqueous alteration on the Moon in the polar regions
- Heat + ice + pressure + long duration = alteration
- Experimental methods to simulate alterations

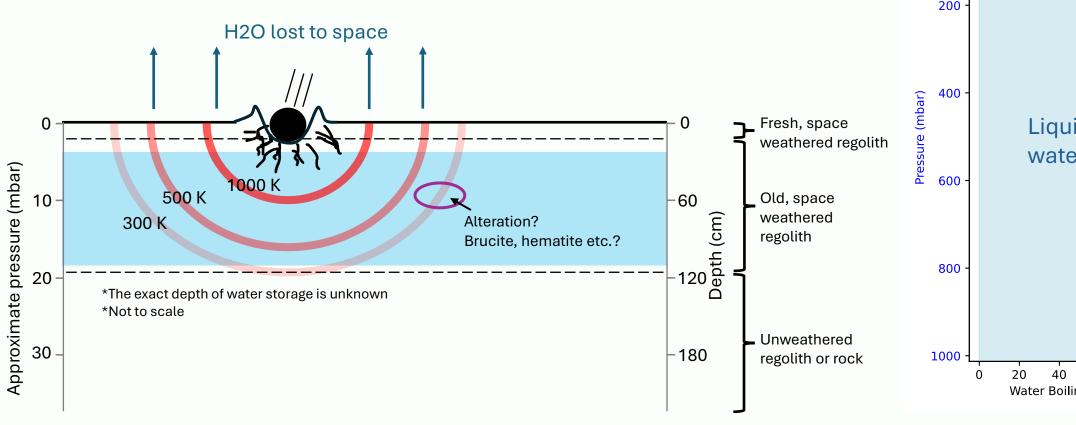


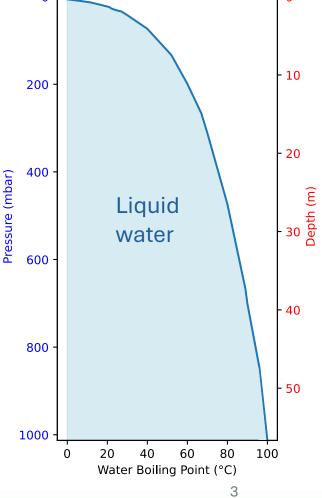
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Experimental methods to simulate alterations





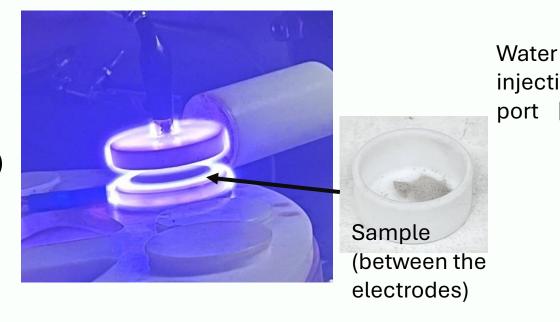
### Experiment 1: Vacuum chamber + Energetic electron impact (EEI)

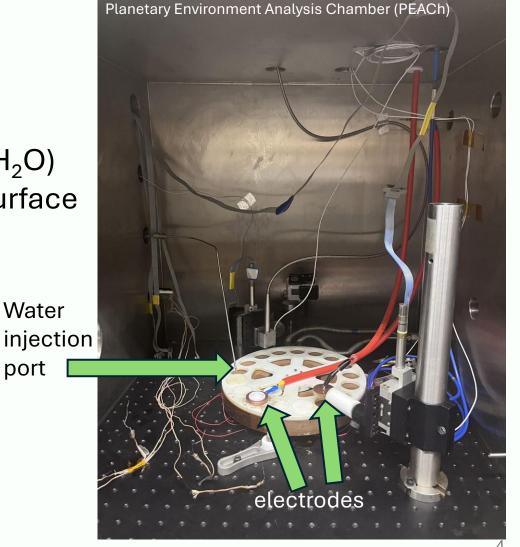
#### **Key functions:**

- Using EEI to activate the sample surface
- React water with olivine and basaltic glass
- Use D<sub>2</sub>O (to distinguish from atmospheric H<sub>2</sub>O)
- Controlled P-T condition ~16m under the surface

Exp 1: Vacuum chamber + EEI

Energetic electron impact (EEI) in Argon





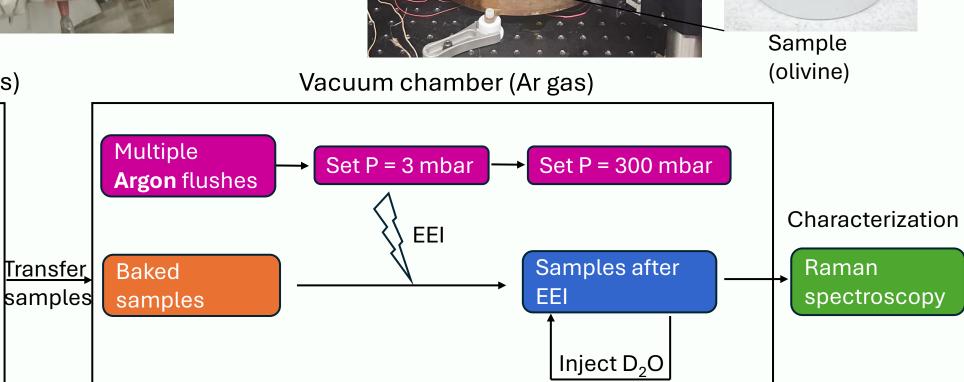
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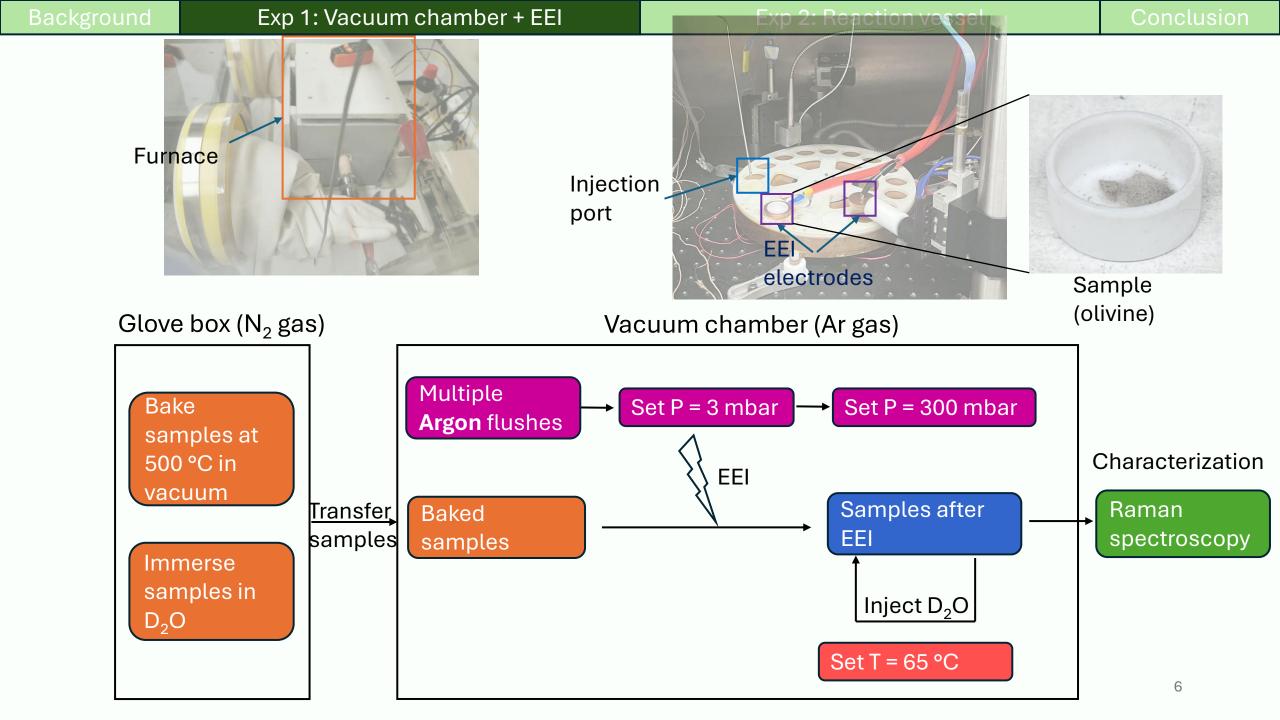


Bake samples at 500 °C in vacuum

Immerse samples in  $D_2O$ 

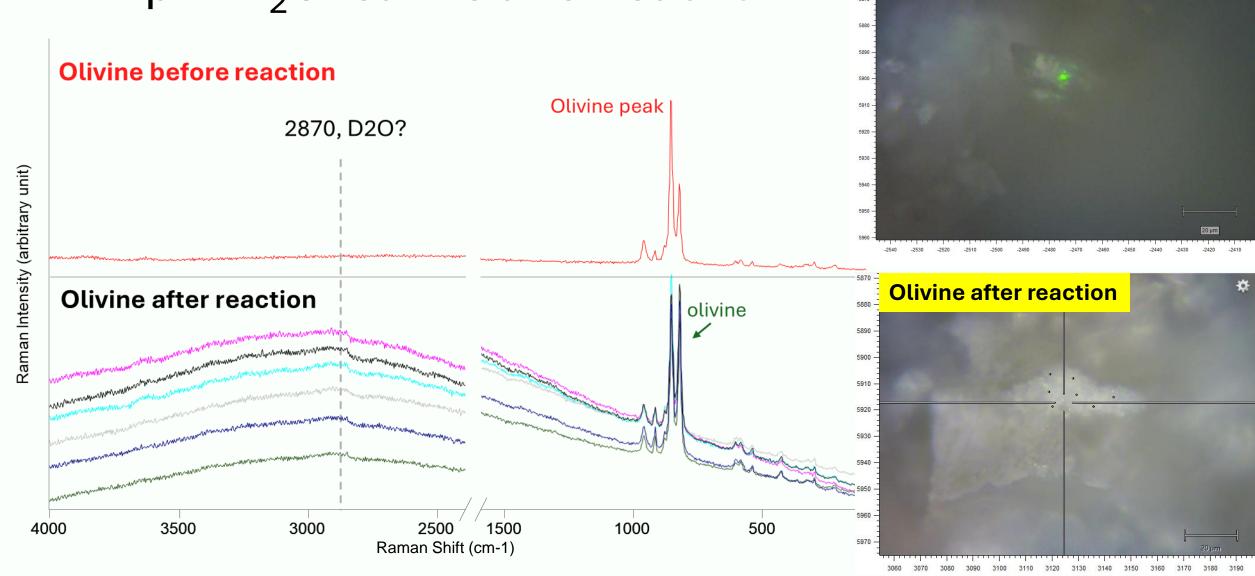


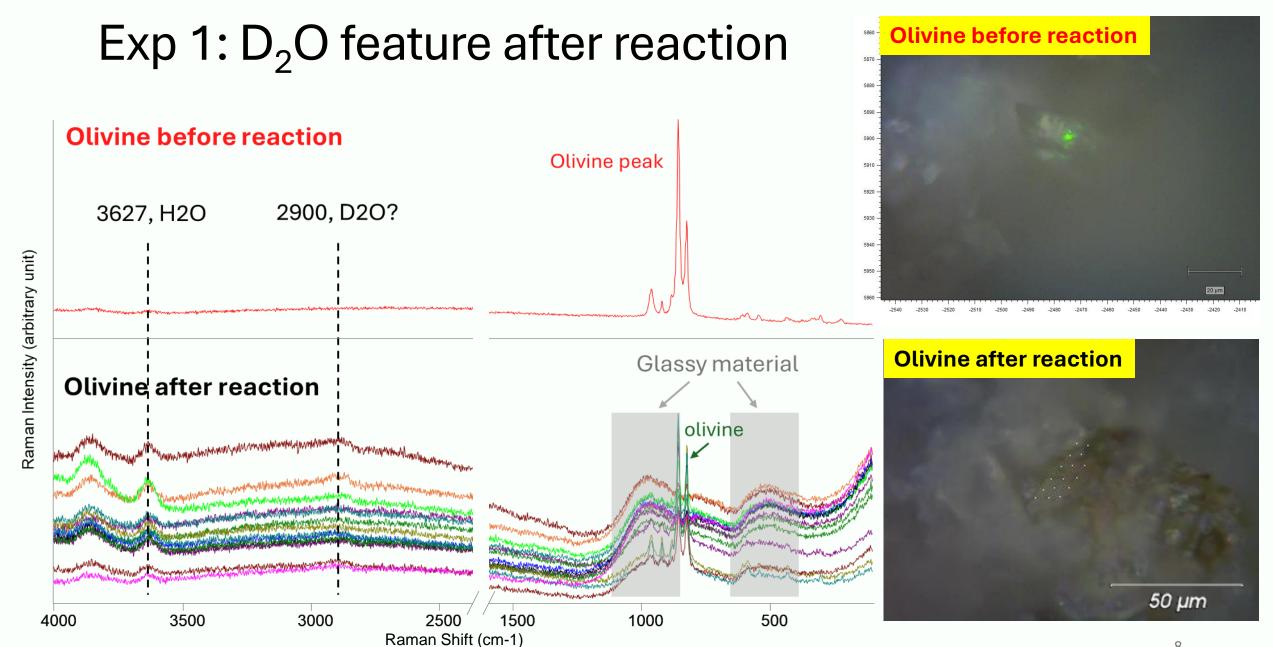
Set T = 65 °C



Olivine before reaction

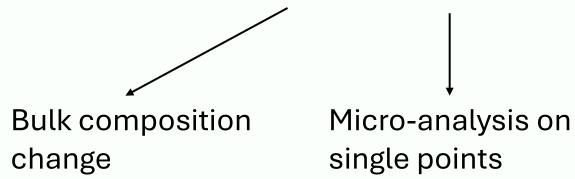






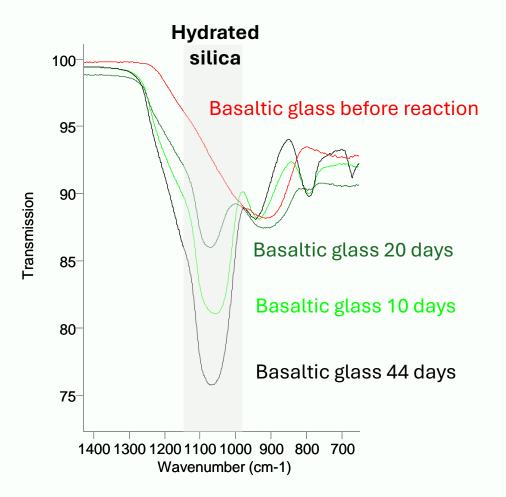


- Teflon-lined
- Solid: olivine or basaltic glass 50 mg
- Solution:  $0.5 \text{ ml } D_2O + 0.5 \text{ ml dilute } H_2SO_4 (16\%)$
- Temperature: 90 °C
- Duration: 10 days, 20 days, 44 days
- Characterization: FT-IR and Raman

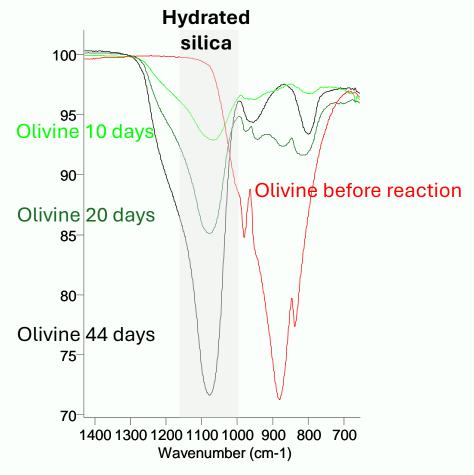


## Exp 2: FT-IR spectra – the main reaction product is hydrated silica

#### Basaltic glass IR spectra

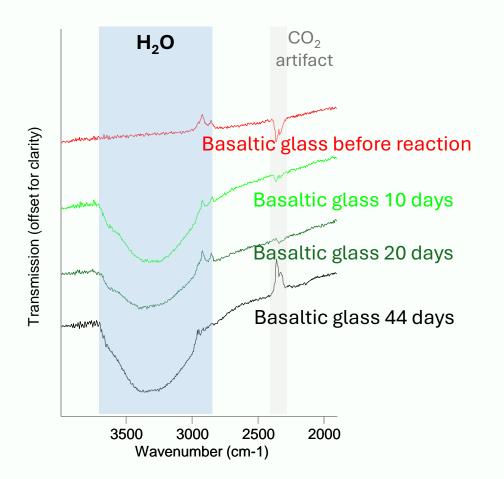


#### Olivine IR spectra

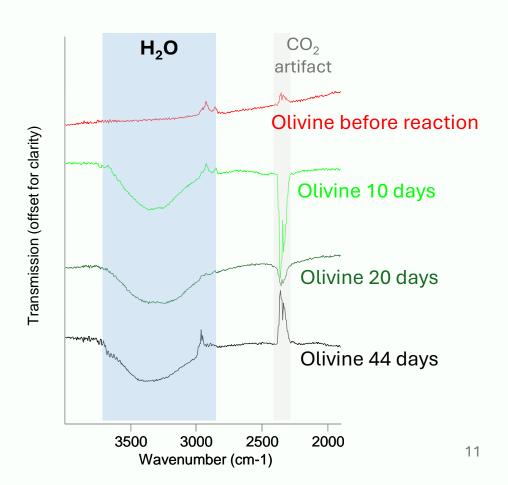


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Basaltic glass IR spectra



#### Olivine IR spectra



Duration	Olivine	Basaltic glass
10 days	Sulfate feature	No apparent change
20 days	Hematite feature	No apparent change
44 days	Completely dissolved	Sulfate+D <sub>2</sub> O+H <sub>2</sub> O

- Olivine: different reaction products depending on reaction time.
- Basaltic glass: dissolves slower than olivine

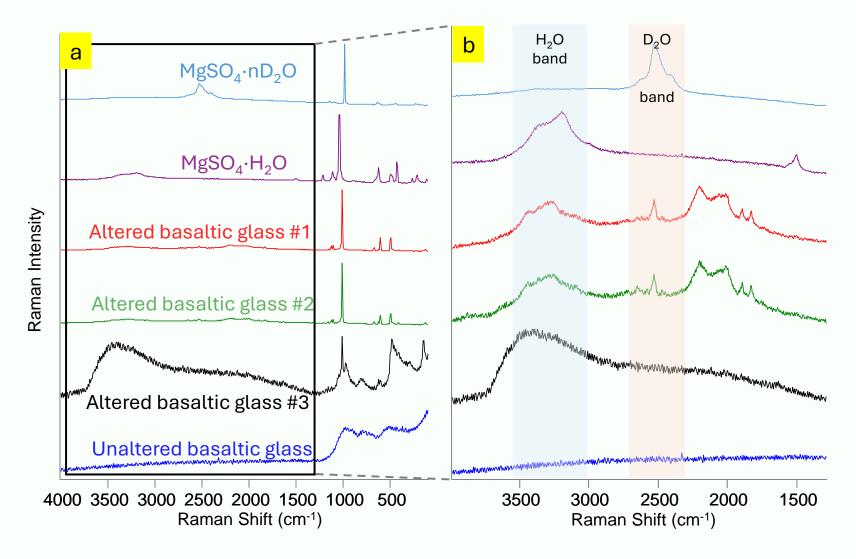
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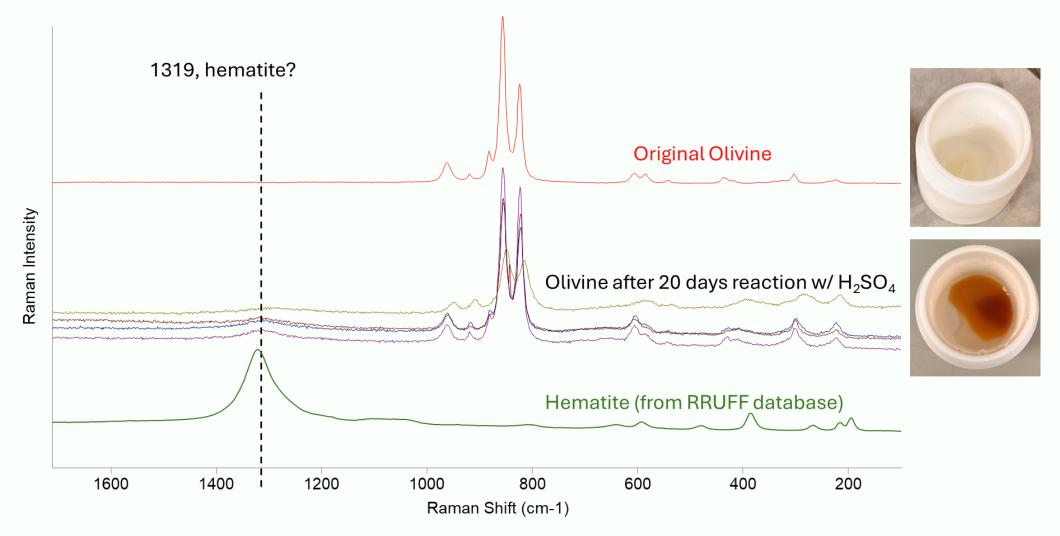
## Exp 2: H<sub>2</sub>O, D<sub>2</sub>O, sulfate Raman features after 44 days of reaction with basaltic glass



Duration	Olivine	Basaltic glass
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44 days	Completely dissolved	Sulfate+D <sub>2</sub> O+H <sub>2</sub> O

- Olivine: different reaction products depending on reaction time.
- Basaltic glass: dissolves slower than olivine

## Exp 2: Hematite Raman feature after 20 days of reaction in olivine



### Conclusions

Exp 1 (EEI +  $D_2O$  + heating)

 D<sub>2</sub>O feature identified by Raman spectroscopy



EEI effectively activates grain surfaces as we might expect occurs on the lunar surface by space weathering

Exp 2 (Acid +  $D_2O$  + longer term heating)



- Dissolution of olivine/basaltic glass
- Evaporation of "brine" produces secondary deposits of salts

## Thank you!

Contact:

Chengzheng@wustl.edu